

Link to important sites

Leading you to learn more

Different areas

- General Chemistry
- Inorganic Chemistry
- Physical Chemistry
- Organic Chemistry
- Industrial Chemistry
- Forensic Chemistry
- Nuclear chemistry
- Chemistry of Metalloids
- Biochemistry

Vision learning website

- **Please Visit the Vision learning Library to view the different areas you need to learn in Chemistry**

- <http://visionlearning.com/library/>

Remember:

Elements give the basis for all these areas

Chemistry

- [CHE-1.1](#) Matter: Atoms, Molecules & States of Matter
- [CHE-1.2](#) Atomic Theory I: The Early Days
- [CHE-1.3](#) Atomic Theory II: Ions, Isotopes and Electron Shells
- [CHE-1.4](#) The Periodic Table of Elements
- [CHE-1.5](#) The Mole: Its History and Use
- [CHE-1.6](#) Chemical Reactions
- [CHE-1.7](#) Chemical Bonding
- [CHE-1.8](#) Chemical Equations
- [CHE-2.1](#) Water: Properties and Behavior
- [CHE-2.2](#) Acids & Bases: An Introduction
- [CHE-2.3](#) Nuclear Chemistry: An Introduction
- [CHE-2.4](#) Organic Chemistry: An Introduction
- [CHE-2.5](#) Carbohydrates
- [CHE-2.6](#) Fats and Proteins

Vision learning website continued....



- Matter
- http://www.visionlearning.com/library/module_viewer.php?mid=49&l=&c3=
- Atomic Theory I
- http://www.visionlearning.com/library/module_viewer.php?mid=50
- Atomic theory II
- http://www.visionlearning.com/library/module_viewer.php?mid=51
- Moles
- http://www.visionlearning.com/library/module_viewer.php?c3=&mid=53&l=
- Nuclear reactions
- <http://www.nde-ed.org/EducationResources/HighSchool/Radiography/nuclearreactions.htm>

Links made in different pages

Introduction page

<http://hyperphysics.phy-astr.gsu.edu/hbase/solids/sili.html>

<http://www.chemguide.co.uk/atoms/propsmenu.html#top>

<http://www.chemguide.co.uk/atoms/properties/ies.html>

http://van.hep.uiuc.edu/van/qa/section/New_and_Exciting_Physics/What_Atoms_Look_Like/

http://www.visionlearning.com/library/module_viewer.php?mid=55

http://www.chem.vt.edu/RVGS/ACT/notes/activity_series.html

Radio activity page

<http://www.visionlearning.com/>

<http://www2.slac.stanford.edu/vvc/theory/nuclearstability.html>

<http://hyperphysics.phy-astr.gsu.edu/hbase/nuclear/radact.html>

http://www.visionlearning.com/library/module_viewer.php?mid=51

Go to Biology page

<http://www.accessexcellence.org/RC/VL/GG/structure.html>

www.webelements.com/webelements/elements/text/Co/key.html

<http://www.mgwater.com/index.shtml>

Electrode potentials page

- http://en.wikipedia.org/wiki/Electrode_potential
- <http://www.chemguide.co.uk/physical/redoxeqia/introduction.html#top>
- <http://members.aol.com/logan20/voltaic.html>
- <http://www.physchem.co.za/Redox/SEP.htm>
- <http://hyperphysics.phy-astr.gsu.edu/hbase/tables/electpot.html#c1>
- <http://hyperphysics.phy-astr.gsu.edu/hbase/chemical/electrochem.html#c1>
- <http://www.eduseek.com/navigate.php?ID=2997>

Activity-series page

- http://www.chem.vt.edu/RVGS/ACT/notes/activity_series.html
- <http://www.harpercollege.edu/tm-ps/chm/100/dgodambe/thedisk/series/series.htm>
- <http://www.harpercollege.edu/tm-ps/chm/100/dgodambe/thedisk/series/3perform.htm>
- http://www.s-cool.co.uk/topic_principles.asp?loc=pr&topic_id=9&subject_id=21

Range of areas in Chemistry

<http://www.101science.com/Chemistry.htm>

<http://www.chem4kids.com/files/elements/index.html>

Data links page

<http://www.chemsoc.org/networks/learnnet/miscon2.htm>

<http://www.chemsoc.org/networks/learnnet/key-subjects.htm>

http://www.chemsoc.org/viselements/pages/data/intro_hydrogen_data.html

http://www.chemsoc.org/viselements/pages/data/intro_groupv_data.html

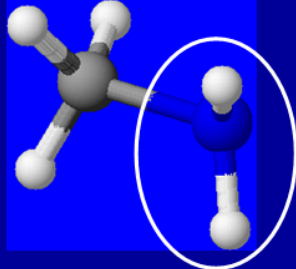
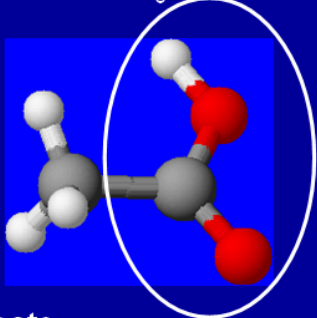
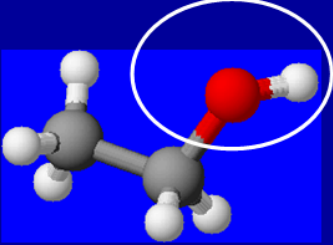
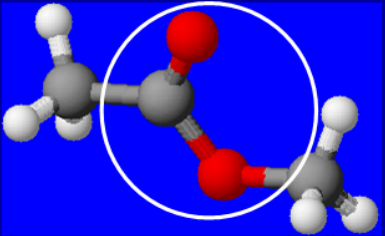
PP slide From Royal Society of Chemistry

visit: <http://www.chemsoc.org/networks/learnnet/cfb/basicchemistry.htm>

This is part of Organic Chemistry, which I have not touched in this website

RSOC
THE INORGANICAL & PHYSICAL CHEMICAL SOCIETY
Founded 1911

Some important functional groups

<p>Methylamine Molecular formula: CH_3NH_2</p> <p>amine group, - NH_2</p> 	<p>Ethanoic acid Molecular formula: CH_3COOH</p> <p>carboxylic acid group - COOH</p> 
<p>Ethanol Molecular formula: $\text{C}_2\text{H}_5\text{OH}$</p> <p>alcohol group, - OH also called a hydroxyl group</p> 	<p>Methyl ethanoate Molecular formula: $\text{CH}_3\text{COOCH}_3$</p> <p>ester group, - COO -</p> 

[Replay](#) [Close window](#)

These functional groups are very important in organic chemistry

And Biology too

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